

Surface-modified carbon nanotubes electrocatalysts for enhanced Lithium-Sulfur Battery performance

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Lithium-sulfur (Li-S) batteries, with a high theoretical energy density of 2600 Wh kg⁻¹, are promising for future energy storage [1]. However, the multiphase sulfur redox reaction remains challenging [2]. Herein, an organic framework-based electrocatalytic polymer composite, incorporating zinc(II) 5,10,15,20-tetrakis-(2,2'-bithien-5-yl)porphyrin (Zn-Por) into polybisthiophene (PBTH) linkers on carbon nanotube (CNT) surfaces, enhances redox kinetics. ZnPor-PBTH/CNTs was characterized by SEM, EDAX, XPS, and FTIR. In Li-S coin cells, electrochemical studies (CV, EIS, GCPL) showed a discharge capacity of 1144.14 mAh g⁻¹ at C/20, low resistance (20 Ω), and stable cycling up to 50 cycles at C/10 with reduced polarization and suppressed polysulfide shuttling.

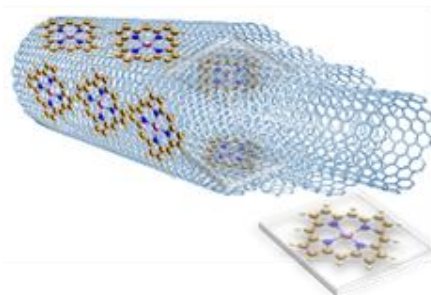


Figure 1. Carbon nanotubes functionalized with Zn-Por sites linked by PBTH.

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References

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